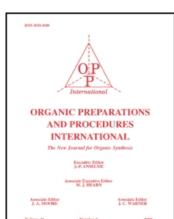
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## 1,1'-DIACETYLFERROCENE MONOTOSYLHYDRAZONE AND 1,1'-DIACETYLFERROCENE DITOSYLHYDRAZONE

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# 1,1'-DIACETYLFERROCENE MONOTOSYLHYDRAZONE AND 1,1'-DIACETYLFERROCENE DITOSYLHYDRAZONE

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Tos: p-CH3C6H4SO2

We wish to report the preparation of 1,1'-diacetylferrocene monotosylhydrazone and 1,1'-diacetylferrocene ditosylhydrazone. These compounds may prove to be valuable precursors to a number of novel ferrocenes.

A mixture of the mono- and ditosylhydrazones was prepared by reaction of 1,1'-diacetylferrocene with 1.5 equivalents of p-toluenesulfonylhydrazine in ethanol. The two products were easily separated by taking advantage of their solubility differences in chloroform-ethanol. The ditosylhydrazone can be obtained as the sole product by using a 1:2 mole ratio of 1,1'-diacetylferrocene to p-toluenesulfonylhydrazine.

### EXPERIMENTAL

1,1 -Diacetylferrocene Monotosylhydrazone and 1,1 -Diacetylferrocene Ditosylhydrazone. To a solution of 2.79 g. (15 mmoles) of p-toluenesulfonylhydrazine in 13 ml. of warm absolute ethanol was added a warm solution of 2.70 g. (10 mmoles) of 1,1 -diacetylferrocene 1 in 35-40 ml. of absolute ethanol. The mixture was refluxed for 30 minutes, cooled to room temperature and filtered to give 4.70 g. of an orange-brown solid. The solid was treated with 60-70 ml. of a hot chloroform-ethanol (7:3) solution, filtered and the filtrate allowed to cool to room temperature whereupon the monotosylhydrazone precipitated. On filtration, 1,1'diacetylferrocene monotosylhydrazone (1.11 g., 25%) was obtained as an orange-brown powder, m.p. 196-199° (dec.). The following infrared spectral bands (KBr disc) are noteworthy: 3275 [w, vNH], 1645 [s, vCO and vC=N], 1609, 1598, 1493 [w-m, phenyl nucleus], 1338 [s,  $v_{AS}SO_2$ ], 1173, 1168 [s, doublet,  $v_sSO_2$ ], 1117 [m, acetyl group], 892, 879 [m, C-H bending mode on ferrocene ring] and 817 cm-1 [m, out-of-plane bending for two adjacent hydrogens on the benzene] (w = weak, m = medium, s = strong). The nmr spectrum (dimethylsulfoxide-d6) showed the following bands: 2.07 [singlet,  $CH_3C=N^-$ ], 2.20 [singlet,  $CH_3CO^-$ ], 2.47 [singlet,  $p-CH_3C_6H_4^-$ ], 3.47 [broad singlet, NH], 4.45 [unsymmetrical triplet, H3.4 of cyclopentadienyl rings], 4.70 [unsymmetrical triplet,  $H_{2.5}$  of cyclopentadienyl rings], 7.60 and 8.03 ppm [AB quartet ( $J_{AB}$  = 25 cps) assigned to the four protons meta and ortho, respectively, to the sulfonyl group].

Anal. Calcd. for  $C_{21}H_{22}FeN_2O_3S$ : C, 57.54; H, 5.06; N, 6.39; S, 7.31. Found: C, 57.29; H, 5.11; N, 6.32; S, 7.47.

The orange, chloroform-ethanol insoluble solid was identified as 1, 1'-diacetylferrocene ditosylhydrazone (2.72 g., 48%) which decomposed above 203°. The following infrared spectral bands (KBr disc) are note-

worthy: 3260 [w-m, vNH], 1648 [w-m, vC=N], 1600, 1496, 1487 [m, phenyl nucleus], 1330 [s,  $v_{AS}$ SO2], 1168, 1161 [s, doublet,  $v_{S}$ SO2], 886 [m-s, C-H bending mode on ferrocene ring] and 815 cm<sup>-1</sup> [m-s, out-of-plane bending for two adjacent hydrogens on the benzene ring]. The ditosylhydrazone was insufficiently soluble in common nmr solvents (CDCl<sub>3</sub>, DMSO-d<sub>6</sub>, C<sub>6</sub>H<sub>6</sub>-d<sub>6</sub>) to obtain a satisfactory nmr spectrum.

Anal. Calcd. for  $C_{28}H_{30}FeN_{4}S_{2}O_{4}$ : C, 55.44; H, 4.98; N, 9.23; S, 10.57. Found: C, 55.39; H, 4.99; N, 9.44; S, 10.28.

1,1'-Diacetylferrocene Ditosylhydrazone. To a solution of 1.87 g. (10 mmoles) of p-toluenesulfonylhydrazine in 8.5 ml. of warm absolute ethanol was added a warm solution of 1.35 g. (5 mmoles) of 1,1'-diacetylferrocene in 18 ml. of absolute ethanol. The mixture was refluxed for a few minutes, cooled to room temperature and then allowed to stand overnight. Filtration gave 2.65 g. (87%) of the ditosylhydrazone.

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